#### 1 **ABSTRACT** 2 Gaseous emissions from swine manure storage systems represent a concern to air quality due 3 to the potential impacts of hydrogen sulfide, ammonia, methane, and volatile organic compounds 4 on environmental quality, human health, and wellbeing. The lack of knowledge concerning 5 functional aspects of swine manure management systems has been a major obstacle in the 6 development and optimization of emission abatement technologies for these point sources. In 7 this study, a classification system based on gas emission characteristics and effluent 8 concentrations of total phosphorus (P) and total sulfur (S) was devised and tested on 29 swine 9 manure management systems in Iowa, Oklahoma, and North Carolina in an effort to elucidate 10 functional characteristics of these systems. Four swine manure management system classes were 11 identified that differed in effluent concentrations of P and S, methane (CH<sub>4</sub>) emission rate, odor 12 intensity, and air concentration of volatile organic compounds (VOCs). Odor intensity and the 13 concentration of VOCs in air emitted from swine manure management systems were strongly correlated ( $r^2 = 0.88$ ). The concentration of VOC in air samples was highest with outdoor swine 14 15 manure management systems that received a high input of volatile solids (Type 2). These 16 systems were also shown to have the highest odor intensity levels. The emission rate for VOCs 17 and the odor intensity associated with swine manure management systems were inversely 18 correlated with CH<sub>4</sub> and ammonia (NH<sub>3</sub>) emission rates. The emission rate of CH<sub>4</sub>, NH<sub>3</sub>, and 19 VOCs were found to be dependent upon manure loading rate and were indirectly influenced by 20 animal numbers. 21 22 Key Words: Odor, Swine Production, Manure Management, Volatile Organic Compounds, 23 Concentrated Animal Feeding Operations. 24 25 **Abbreviations**: FID, Flame ionization detector; GC, Gas chromatography; ICP-AES, Inductively-coupled plasma-atomic emission spectrometry; TD, Thermal desorption; VOC, 26

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Volatile Organic Compound.

#### 1 **INTRODUCTION** 2 3 In the United States, approximately 157,000 swine production facilities produce 4 103,000,000 hogs yr<sup>-1</sup> that are marketed in the U.S. and globally (ASAE, 1988; Harkin, 1997). Annually, these production facilities produce more than 116,652,300 tons yr<sup>-1</sup> of manure that is 5 6 often stored for periods up to 13 months before land application (ASAE, 1988; Harkin, 1997). 7 Air quality studies have indicated that emissions released from stored swine manure have the 8 potential to decay local, regional, and global air quality through the discharge of ammonia (NH<sub>3</sub>) 9 (Harper and Sharpe, 1997; Asman, 1995), nitrous oxide (N<sub>2</sub>O) (Sharpe and Harper, 1998), 10 methane (CH<sub>4</sub>) (Safley et al., 1992; Sharpe and Harper, 1997), hydrogen sulfide (H<sub>2</sub>S) (Jacobson 11 et al., 1997a), particulate matter (VanWicklen, 1997), and volatile organic compounds (VOCs) 12 (Zahn et al., 1997). Much research has focused on the development of strategies to reduce or 13 eliminate emissions and odors from stored animal manure. However, a major part of this 14 research has not been applied by the swine industry due to economic restrictions or due to 15 sporadic or ineffective performance of emission abatement approaches (Miner, 1982; Miner, 16 1995; Miner, 1999). Performance evaluations of emission abatement strategies often cite poorly 17 understood microbiological processes or other poorly defined intrinsic properties of swine 18 manure management systems as the reason for ineffective performance of a particular emission 19 abatement method (Miner, 1995). However, the exact cause for many of these failures has 20 remained speculative due to the lack knowledge concerning functional aspects of animal manure 21 management systems. 22 The purpose of this research was to develop a method to functionally classify swine 23

The purpose of this research was to develop a method to functionally classify swine manure management systems based on effluent chemical properties and emission rates of CH<sub>4</sub>, NH<sub>3</sub>, H<sub>2</sub>S, and VOC. A reliable method to functionally classify manure management systems would serve the agricultural industry as a management tool in evaluating best management practices for swine manure storage systems and would serve regulators as a rapid method to identify production sites that represent a potential air quality or nuisance concern.

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1 2	MATERIALS AND METHODS
3	Physical and Chemical Characterization of Stored Manure
4	Effluent samples (200 ml each) and pH were taken at six locations at the center and around
5	the perimeter of the manure management system at a minimum distance of 2.5 meters from the
6	edge of the storage impoundment and at a depth of 2 cm according to the method of DiSpirito et
7	al., (1995). Measurements of pH were taken with a portable pH/temperature meter (Model
8	#59002-00, Cole Parmer, Vernon Hills, IL). Total carbon (C), nitrogen (N), and hydrogen (H)
9	were determined on oven (100° C) dried effluent samples using a Model 2100 Perkin Elmer
10	CHN analyzer. Percent values were converted to g L <sup>-1</sup> values based on solids content determined
11	by gravimetric methods. Volatile solids concentration was determined by the difference in
12	weight of oven-dried (100° C) and ashed (550° C) samples. Analysis of main-group elements
13	and transition metal cations was performed on microwave digested (method SW 846-3015, CEM
14	Corp., 1996) effluent samples according to EPA method SW 846-3015 (CEM Corp., 1996).
15	Quantitative analysis of digested samples was performed using a Thermo Jarrel Ash Model ICAF
16	61E Inductively Coupled Plasma-Atomic Emission Spectrometer (ICP-AES) and elemental
17	concentration is reported as the mean $\pm$ the standard error. Other physical and chemical
18	parameters evaluated in this study included bacterial chlorophyll $a$ and $b$ concentration.
19	Bacteriochlorophyll $a$ and $b$ was determined as previously described by Siefert et al. (1978) and
20	DiSpirito et al. (1995). The bacteriochlorophyll a concentration was calculated
21	spectrophotometrically using a molar absorptivity coefficient of $\varepsilon_{777} = 75 \text{ cm}^{-1} \text{ mM}^{-1}$ . Lagoons
22	with bacteriochlorophyll $a$ concentrations above 40 nmol/ml are subsequently referred to as
23	"photosynthetic lagoons".
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25	Description of Swine Manure Storage Facilities, Placement of Air Monitoring Equipment,
26	and Statistical Analyses
27	Loading criteria and individual site descriptions for 29 swine manure management systems
28	located in Iowa ( $n = 24$ ), Oklahoma ( $n = 2$ ), and North Carolina ( $n = 3$ ) that were sampled during
29	the months of August and September 1997 are described in Table 1. Individual manure
30	management systems were separated into four main categories (Type 1, Type 2, Type 3, and
31	Type 4) based on the concentration of total P and total S present in effluent samples. Average

1 physical and chemical properties, as well as management attributes for these systems are

2 summarized in Table 1. Meteorological conditions (wind speed, relative humidity, irradiance,

3 solution temperature at a 10 cm depth, and air temperature) were monitored continuously and the

sample mean was recorded in 0.5 minute intervals at the point of air sample collection by an

integrated weather station (Sauer and Hatfield, 1994) that was positioned at the center of the

outdoor manure storage systems. Height of the sensors was established by trajectory simulation

models described in the micrometeorological flux methods section (theoretical profile shape) and

was identical to the air sampling height used for micrometeorological flux measurements.

Micrometeorological data was not collected for studies that employed dynamic flux chamber methods for determination of CH<sub>4</sub> flux rates.

Statistical evaluation of data and experimental designs were performed with JMP version 3 statistical discovery software (SAS Institute, Inc., Cary, NC).

#### Capture and Analysis of Air Pollutants from Swine Manure Management Systems

Volatile organic compounds (VOC) were captured on a multibed adsorbent tube containing a combination of Tenax TA and Carboxen-569 (Supelco, Bellefonte, PA), according to the low-volume sampling method developed by Zahn et al. (1997). For air sampling of VOC, flow rate through individual thermal desorption tubes was precisely regulated at 1.0 L min<sup>-1</sup> throughout the 30 min sampling period, using thermal mass flow controllers (Series 810, Sierra Instruments, Inc., Monterey, CA) that were connected to a common, high-vacuum manifold. Desorption tubes were analyzed by gas chromatography using a flame ionization or mass selective detector as previously described by Zahn et al. (1997).

Ammonia (NH<sub>3</sub>) was collected from air using two glass impingers (210 ml internal volume) arranged in series, each containing 25 grams of 2 mm glass balls. Air samples were drawn by vacuum through a submerged fritted glass diffusion tube into 60 ml of 0.2M boric acid using a Buck IH sampling pump operated at 1.0 L min<sup>-1</sup>. The boric acid solution was replaced in 30 to 120 minute intervals depending on proximity of the sampler to the emission source. Ammonium concentration was determined by the salicylate-nitroprusside technique according to U.S. EPA method 351.2 (U.S. EPA, 1979). Solutions of ammonium chloride in 0.2M boric acid were utilized as reference standards to determine ammonium ion concentration. Hydrogen sulfide

by previously described gas chromatographic methods using either a 0.32 mm x 30 m SPB-1 sulfur fused silica column with flame photometric detection (Bulletin 876, Supelco, Inc., Bellefone, PA) or a 1/8" x 8' HayeSep-Q packed column with thermal conductivity detection (Chan et al., 1998), respectively. Air samples for H<sub>2</sub>S and CH<sub>4</sub> analysis were drawn by vacuum into 1.0 L Tedlar gas sampling bags using a model 1062 grab sampler (Supelco, Bellefonte, PA) operated at 25 mL min<sup>-1</sup>. Teflon surfaces were equilibrated with analytes (H<sub>2</sub>S and CH<sub>4</sub>) present in the air sample by performing four fill-purge cycles before collection of the final air sample. Lead acetate strips (Model #701, VICI Metronics Inc., Santa Clara, CA) were utilized at the point of air sample collection (by Tedlar bag methods) to confirm gas chromatographic results for the quantification of H<sub>2</sub>S. The latter methods indicated that H<sub>2</sub>S losses due to reaction or

(H<sub>2</sub>S) and CH<sub>4</sub> were collected in 1.0 L Tedlar gas sampling bags and evaluated in the laboratory

of air sample collection (by Tedlar bag methods) to confirm gas chromatographic results for the quantification of H<sub>2</sub>S. The latter methods indicated that H<sub>2</sub>S losses due to reaction or condensation were less than 4% if samples were analyzed within 6 hours of collection. The concentration of CH<sub>4</sub> in air was also determined in real-time at two of the 29 swine manure management systems noted in Fig. 1 using a tunable-diode laser system operated in the infrared region at 2968.4034 cm<sup>-1</sup>, as previously described by Simpson et al., (1995). Concentration measurements of CH<sub>4</sub> that were determined using the tunable-diode laser were converted to flux using the theoretical profile shape micrometeorological method. All other CH<sub>4</sub> flux measurements performed in this study used CH<sub>4</sub> concentration measurements that were determined by the gas chromatographic method. A combination of the air sampling methods described above were used to determine recovery efficiency and to validate the H<sub>2</sub>S and CH<sub>4</sub>

sampling techniques.

#### **Evaluation of the Odor Intensity Associated with Swine Manure Management Systems**

Odor intensity was measured by the method of direct scaling in reference to an odor standard of defined intensity using three to four trained panelists (Cain et al., 1998; Degel and Koster, 1998; Liden et al., 1998; Livermore and Laing, 1998). Direct scaling was based on estimation of the intensity of olfactory sensations associated with an odor source by assigning numerical values to sensory stimuli. Sensory responses were normalized against the artificial swine odor reference standard "Z-2" (Zahn and DiSpirito, 1999), that consisted of: 0.05 mM dimethyl disulfide, 8 mM acetic acid, 3.5 mM propionic acid, 0.5 mM isobutyric acid, 0.4 mM 2-butanol, 1.4 mM butyric

1 acid, 0.2 mM isovaleric acid, 0.5 mM valeric acid, 0.1 mM isocaproic acid, 0.2 mM caproic acid, 2 0.2 mM heptanoic acid, 0.1 mM indole, 0.15 mM 3-methyl indole, 0.2 mM 4-methyl phenol, 3 0.12 mM 4-ethyl phenol, 0.15 mM phenol, 0.1 mM benzyl alcohol, 0.15 mM 2-amino 4 acetophenone, 0.1 mM butylated hydroxytoluene (added as a preservative), and 8 mM 5 ammonium acetate. Chemical composition of the artificial swine odor Z-2 was optimized in a 6 laboratory dynamic flux chamber to mimic emission parameters for VOCs emitted from a 7 manure sample collected from a high-odor, Type 1 swine manure management system (Zahn et 8 al., 2000b). Pure compounds were dissolved in warm (45° C) water while stirring and the 9 solution pH was frequently adjusted to pH 7.0 with 2 M potassium hydroxide. Approximately 15 10 mL of the solution was transferred to an amber serum vial (30 ml nominal volume), was capped 11 with a silicone/Teflon septum, and then degassed under repeated cycles of vacuum and argon to 12 create an anaerobic headspace. Samples were stored in the dark at ambient temperatures until 13 they were used. Panelists were provided the synthetic swine odor solution Z-2 in an uncapped 30 14 ml serum vial, a solution of 2 mM n-butanol in a 30 ml serum vial, and a site odor evaluation 15 worksheet, previously described by Zahn (1997). Panelists then evaluated the n-butanol solution 16 that was considered neutral (grade 3/10), and the Z-2 solution that was considered unpleasant 17 (grade 6.5/10), at a neutral distance (>1000 m) from the manure management system. Panelists 18 were then positioned in the emission plume from the manure management system (1.5 m from 19 the emitting source at a height of 1.5 m) and asked to compare the odor intensity to the reference 20 standards. Numerical evaluations of the swine manure management systems ranged from neutral 21 (3) to unbearable (10), and are reported as the sample mean. Air samples for VOC analysis were 22 collected at the receptor (1.5 m from the emitting source at a height of 1.5 m) throughout the 23 odor evaluation period.

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# Measurement of CH<sub>4</sub> Flux, VOC Concentration, and Odor Intensity from 29 Swine Manure Management Systems

An initial screening study was completed at 29 swine manure management systems to measure the flux rate of CH<sub>4</sub>, odor intensity, concentration of VOCs in air at the point of the odor intensity measurements, and analytes in the effluent fraction. The flux rate of CH<sub>4</sub>, for this initial screening study, was measured at 29 manure management systems (lagoons, earthen basins,

1 cement-lined basins, steel-lined tanks, and confinement pits) using a flux chamber described by 2 DiSpirito and Zahn (1998). The dynamic flux chamber method was found to be most suitable for 3 screening large sample numbers due to the minimum operation requirements, portability, and 4 reliability for measurement of non-reactive gases such as CH<sub>4</sub> (Chan et al., 1998). The dynamic 5 flux chamber was positioned near the center of the swine manure management system, with the 6 exception of deep and shallow pit systems, which were sampled at the pump-out positions near 7 the wall of the pit. Chamber operational parameters were modified from semi-static to dynamic 8 mode by installation of a sweep gas manifold. Compressed air (containing 1.1 parts per million by volume or 0.77 mg m<sup>-3</sup> CH<sub>4</sub>) sweep gas was provided to the enclosure at a flow rate of 2.0 L 9 min<sup>-1</sup> in a demand mode. Air pressure within the chamber was maintained at barometric pressure 10 11 through the use of a silicone oil (impinger) purge valve, positioned on the gas supply manifold. 12 The static pressure differential was maintained at  $0 \pm 1.5$  kPa throughout the collection period 13 with the use of a mercury manometer. Gases were removed from the chamber through 0.635 cm 14 ID Teflon tubing that was attached to a vacuum pump through an inline mass flow controller set 15 at 2.0 L min<sup>-1</sup>, following a 1 hr chamber equilibration period. Six individual air samples (~750 16 ml each) were collected over a 3-hour sampling period. Samples were analyzed for CH<sub>4</sub> 17 concentration by gas chromatography within 6 hours of sample collection. The concentration of 18 CH<sub>4</sub> in air samples was converted to trace gas flux density (f) through the equation: 19 (1)  $f = (s/a) (C_0 - C_i)$  where: s = sweep flow rate (2.025 L min-1), A = chamber basal area (2500 cm<sup>2</sup>),  $C_0$  = concentration of 20 methane in the exit air (mg m<sup>-3</sup>), and  $C_i$  = concentration of CH<sub>4</sub> in the sweep gas (0.77 mg m<sup>-3</sup>). 21 22 In addition to the collection of samples for CH<sub>4</sub> flux measurements, odor intensity and VOC 23 concentration in air were taken at the receptor (1.5 m from the emitting source at a height of 1.5

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#### **Detailed Studies of Gas Flux Rates from Four Swine Manure Management Systems**

m). Samples were collected at 1.0 L min<sup>-1</sup> during the 30 min olfactory evaluation period.

The flux rate of CH<sub>4</sub>, H<sub>2</sub>S, NH<sub>3</sub>, and VOCs was measured continuously at four swine manure management systems, throughout a 24 hour period using the theoretical profile shape micrometeorological flux measurement method. The four sites chosen for this comprehensive study represented each of the four classes of swine manure management systems defined in Table

1 1. The theoretical profile shape method was chosen instead of chamber methods, since the latter

methods adversely affected VOC emission profiles. Higher flux rates of disulfide compounds

(dimethyl disulfide and dimethyl trisulfide) and lower flux rates of sulfide compounds (methyl

mercaptan) were consistently observed with flux chamber methods when compared to

5 micrometeorological methods. This observation indicated that the chamber surfaces were

potentially the site of oxidation reactions for reactive gases (H<sub>2</sub>S, NH<sub>3</sub>, and VOCs). These

results are consistent with previous studies that show significant reduction in the concentration of

VOCs collected from air if samples were drawn through glass or polymeric tubing (Zahn et al.,

9 1997).

The flux of NH<sub>3</sub>, H<sub>2</sub>S, CH<sub>4</sub>, and VOCs from the lagoon were measured using the theoretical profile shape method described by Wilson and coworkers (1982). The emission rate of gases from a circular source plot was calculated with the following equation:

13 (2) 
$$\overline{F}_z(0) = \frac{(uc)_{measured}}{\Phi}$$

Where  $F_z(0)$  is the vertical flux rate in  $\mu g$  cm<sup>-2</sup> s<sup>-1</sup>,  $\Phi$  is the non-dimensional normalized horizontal flux predicted by the trajectory simulation model and  $(uc)^{measured}$  is the product of the measured average wind velocity and air concentration of analyte in m s<sup>-1</sup> and  $\mu g$  m<sup>-3</sup>, respectively (Majewski et al., 1989; Majewski, 1990). Flux measurements were completed at the center of swine waste management systems with a circular shape. System classification based on the concentration ratio of phosphorus to sulfur (Table 1) and the requirement for circular manure management systems served as the major criteria for selecting the four swine waste management systems that were sampled for follow-up studies.

The surface roughness was determined before the sampling period began by performing mean wind velocity profile measurements at 0.2, 0.5, 1.0, 2.0, and 3.0 m for a period of 1 hour at the center of the swine manure management system with cup anemometers (model 03101-5, R.M Young Co., Traverse City, MI). The mean value for roughness length at the surface of outdoor swine manure management systems, during periods of neutral atmospheric stability (midmorning), was  $0.10 \pm 0.02$  cm (mean  $\pm$  std. error). The following measurement parameters were used to establish the sampling height: Type 1 = total mechanically-ventilated exhaust flow rate =  $110,000 \text{ m}^3 \text{ hr}^{-1}$ , sampling position at the fan orifices (4 total); Type 2 = 39 m diameter basin,

1 z=ZINST(0.1 cm roughness length, 1950 cm radius = 70 cm sampling height); Type 3 = 92 m 2 diameter lagoon, z=ZINST(0.1 cm roughness length, 4600 cm radius = 189 cm sampling height); 3 Type 4 = a primary 100 m diameter photosynthetic lagoon, z=ZINST(0.1 cm roughness length, 4 5000 cm radius = 191 cm sampling height). Error in measurement height that was associated 5 with temporal changes in roughness length for the lagoon surface was estimated to cause a 6 maximum error of 8% in the height parameter for emission measurements. 7 Flux measurements using the theoretical profile shape method were based on the 8 concentration of airborne analytes present at a measurement height (z) and meteorological data, 9 collected at the same point (Majewski, 1990). Measurement height (z) was calculated by 10 trajectory simulation models and was based on system surface area and roughness length (Wilson 11 et al., 1982). In addition to the air samples that were collected over the emitting source, 2-5 air 12 samples were collected at the beginning of each sampling period, upwind from each manure 13 storage system, to assess background air concentrations of target analytes and to confirm the 14 source of these emissions. Background analyte concentrations were assumed to remain constant 15 throughout the sampling period and were subtracted from analyte concentrations that were 16 observed above each emitting source. Background concentrations for analytes at each of the four 17 sites sampled in follow-up studies were: Type 1: 6.0 parts per billion by volume (ppbv) (8.5 µg  $m^{-3}$ )  $H_2S$ , 15.0 ppbv (11.1  $\mu$ g  $m^{-3}$ )  $NH_3$ , and 1.3 parts per million by volume (ppmv) (0.91 mg  $m^{-3}$ 18 <sup>3</sup>) CH<sub>4</sub>. Type 2: 15 ppbv (21.3 µg m<sup>-3</sup>) H<sub>2</sub>S, 11.0 ppbv (8.1 µg m<sup>-3</sup>) NH<sub>3</sub>, and 1.3 ppmv (0.91 mg 19 m<sup>-3</sup>) CH<sub>4</sub>. Type 3: 12 ppbv (17.1 μg m<sup>-3</sup>) H<sub>2</sub>S, 7.0 ppbv (5.2 μg m<sup>-3</sup>) NH<sub>3</sub>, and 1.2 ppmv (0.84) 20  $\text{mg m}^{-3}$ ) CH<sub>4</sub>. Type 4: 9.0 ppby (12.8  $\text{ug m}^{-3}$ ) H<sub>2</sub>S, 8.0 ppby (5.9  $\text{ug m}^{-3}$ ) NH<sub>3</sub>, and 1.3 ppmy 21 22 (0.91 mg m<sup>-3</sup>) CH<sub>4</sub>. The concentration of VOCs in background air samples was below the 23 detection limit of 0.2 ppb for all samples that were analyzed. 24 Flux rate measurements for the mechanically-ventilated deep-pit swine manure 25 management system (animal confinements) was performed by continuously monitoring exhaust 26 flow rate during the sample collection period at two pit fan and two ventilation fan orifices using 27 a 3-D sonic anemometer (Campbell Scientific, Logan, UT). Air samples were collected near the

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#### RESULTS AND DISCUSSION

lower sensor arm on the anemometer in the exhaust stream.

## Initial Evidence that Methane Emission Rate is Dependent Upon Manure Management System Environment

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Studies were conducted on two types of manure management systems located on a feederto-finish swine production facility in central Iowa in August 1997 to assess differences in CH<sub>4</sub> emission rate that could be attributed to differences in manure management system environment. Swine waste material entering the manure management system environments originated from animals of the same breeding population that were fed identical diets, were of similar weight, and were managed under similar management routines (feeding schedules and rates). Volatile solids loading rate, however, differed between the two systems by over 300-fold (37 versus 0.12 kg volatile solids day<sup>-1</sup> m<sup>-3</sup>). For the basin system (37 kg volatile solids day<sup>-1</sup> m<sup>-3</sup>), manure was emptied daily from the confinement into an outdoor concrete holding tank and for the lagoon system (0.12 kg volatile solids day<sup>-1</sup> m<sup>-3</sup>), manure was flushed into the earthen holding basin by an intermittent loop flush system. Methane flux measurements were performed on three separate occasions in August 1997 with a tunable-diode laser over the center of the manure management systems using the theoretical-profile-shape method for measuring evaporative fluxes (Fig. 1). While similar CH<sub>4</sub> flux values were observed for systems over the three separate sampling periods, only a single comparison is reported, since this sampling period demonstrated the greatest similarity between sites for environmental conditions known to influence CH<sub>4</sub> flux. Measured environmental conditions over the 68 hour sampling period included wind speed (2.2±0.14 versus 2.4±0.12 m s<sup>-1</sup> [mean and standard error mean]), irradiance (298±35 versus 305±30 W cm<sup>-2</sup>), relative humidity (RH) (85±2 *versus* 78±2 % RH), air temperature (18.6±0.4 versus 22.6±0.5°C) and solution temperature at a 10 cm depth (27±0.2 versus 23±0.3°C) for the lagoon and basin, respectively. Results from CH<sub>4</sub> micrometeorological flux measurements indicated that there were statistically-significant differences in both the flux rate (0.35±0.02) versus 0.13±0.01 μg CH<sub>4</sub> cm<sup>-2</sup> s<sup>-1</sup> [mean and standard error mean]) and emission rate (14.7 versus 0.5 g CH<sub>4</sub> system<sup>-1</sup> s<sup>-1</sup>) for the lagoon and basin, respectively (Fig. 1). Current literature values for CH<sub>4</sub> emissions from stored swine manure have been reported over a range of nearly two orders of magnitude (Harper and Sharpe, 1997). Harper and Sharpe (1997) proposed that the discrepancies between emission values might be explained by differences in measurement methods or due to atypical flux event periods. In addition to these explanations, results described in these experiments indicate that manure management environment, and specifically loading rate, may significantly influence the flux rate of CH<sub>4</sub>. While these observations are not novel in the context of laboratory anaerobic digestion processes (Oleszkiewicz and Sharma, 1990; Hill and Bolte, 1989), they do represent the first report of such a phenomena occurring under production scale conditions in an animal manure management system. The proceeding experiments were focused on defining relationships between effluent chemical properties of various swine manure management systems and observed emission characteristics in order to further define the relationship between system loading rate and emission rate of CH<sub>4</sub>, NH<sub>3</sub>, H<sub>2</sub>S, and VOCs.

### Classification of 29 Swine Manure Management Systems Based on Solution-Phase Chemistry and Methane Emission Rate

The concentration range for elements found in the swine manure storage systems sampled in this study were found to be similar to those previously reported by Giusquiani et al (1998), Japenga and Harmsen (1990), and Zahn et al (1997) (Table 1). There was a weak positive correlation observed between volatile solids loading rate and concentration of elements for each of the samples evaluated (Table 1). Individual correlations between element concentration and volatile solids loading rate indicated that volatile solids loading rate could account for 48 to 76% of the variability observed with effluent concentration of elements. This inadequacy of volatile solids loading rate to account for differences in effluent elemental composition for various swine manure management systems, in addition to the low-throughput of volatile solids measurements, indicated that the solution concentration of elements may provide a more appropriate means to classify swine manure management systems.

In the search for useful effluent chemical classification criteria, 45 pairwise comparisons were made between the concentrations of elements present in the 29 site samples. The nonparametric measure of association for each pairwise comparison is shown in Table 2. In general, a lower level of correlation was observed for pairwise comparisons made between elements showing opposite partitioning behavior (Table 2). For example, transition metal ions have been shown to partition strongly into the sludge or particulate fraction of the manure, while sodium (Na), potassium (K), and to a lesser extent, phosphorus (P), magnesium (Mg), and sulfur

1 (S), demonstrate neutral or preferential partitioning behavior into the supernatant fraction of the 2 manure (Giusquiani et al., 1998; Zahn et al., 1997). Therefore, the concentration of transition 3 metal ions in samples is highly dependent upon the concentration of suspended solids, while Na 4 and K are nearly independent of the concentration of suspended solids. High correlation 5 coefficients were observed for pairwise comparisons between several transition metals, for 6 several transition metals and a small number of main group elements, and finally, for P and S 7 concentrations in the 29 site samples. However, only one pairwise comparison (P and S) 8 demonstrated clear functional clustering of manure management systems based on system CH<sub>4</sub> 9 emission rate, air concentration of VOCs, or odor intensity (Fig. 2, Table 2). 10 The 29 swine manure management systems were observed to cluster into four distinct 11 system subtypes (Type 1, Type 2, Type 3, and Type 4) based on the concentration of S and P in 12 the effluent fraction (Fig. 2, Table 1). No statistical differences were observed for CH<sub>4</sub> emission 13 rates between Type 1 and 2 systems, nor between Type 3 and 4 systems. However, CH<sub>4</sub> emission 14 rates for all other comparisons were statistically different (Fig. 2B). The mean odor intensity 15 values for sampling sites showed a clustering behavior that was similar to the clustering of data 16 points based on effluent P and S concentrations or system CH<sub>4</sub> emission rate. However, odor 17 intensity showed nearly an inverse relationship to CH<sub>4</sub> emission rate, with the highest odor 18 intensities occurring with manure management systems having the lowest CH<sub>4</sub> emission rates 19 (Type 1 and 2 systems). 20 Odor intensities and VOC concentrations in Type 2 systems were consistently higher than 21 measurements performed on Type 1 systems. This result was not entirely consistent with effluent 22 P and S concentrations or with system CH<sub>4</sub> emission rates, which predicted Type 1 systems to 23 have the highest odor intensities and air concentrations of VOCs. Gas transfer coefficients for 24 CH<sub>4</sub> and VOCs are known to differ by several hundred-fold (MacIntyre et al., 1995). In addition 25 to large differences in gas transfer coefficients for CH<sub>4</sub> and VOCs, surface exchange rate for 26 some VOCs is known to be influenced by effluent chemical events that include ionization (pH), 27 hydrogen bonding, and surface slicks (MacIntyre et al., 1995). For these reasons, CH<sub>4</sub> is not an 28 appropriate criterion for predicting VOC volatilization potential between swine manure 29 management systems. Wind, temperature, and irradiance are known to be major factors in the 30 emission rate of sparingly soluble gases (VOCs) from liquid or semi-solid surfaces (MacIntyre et

1 al., 1995; Zahn et al., 1997). Based on the VOC transfer coefficient calculations from MacIntyre 2 et al. (1995), the difference due to wind and temperature exposures between outdoor and indoor 3 manure management systems can account for between 51% to 93% of the observed differences in 4 VOC emissions. This analysis provides evidence that exposure factors can account for 5 differences observed in VOC flux rates, VOC air concentrations, and odor intensity between 6 Type 1 and 2 systems. 7 Swine manure management systems are often designed to release gases produced in 8 anaerobic decomposition processes into the atmosphere. Within these systems, a complex 9 consortium of microorganisms (anaerobic food chain) decompose complex biological waste 10 material to end products including CH<sub>4</sub>, H<sub>2</sub>S, CO<sub>2</sub>, and NH<sub>3</sub> (Gottschalk, 1988; Lana et al., 1998; 11 Mackie et al., 1998; Fenchel and Finlay, 1994). The anaerobic food chain is often functionally 12 separated into microorganisms catalyzing acid-producing reactions from complex organic 13 substrates and Archaea, that catalyze CH<sub>4</sub>-producing reactions from products formed in the 14 breakdown of complex organic substrates (Deppenmeier et al., 1996; Fenchel and Finlay, 1994). 15 The emission rate of CH<sub>4</sub> and partially decomposed microbial substrates (volatile fatty acids) has 16 been previously employed as an indicator to assess functional coupling between processes in the 17 anaerobic food chain in anaerobic digesters (Hill and Bolte, 1989). Overloaded anaerobic 18 digestion processes have been correlated with high emission rates of VOCs and low emission 19 rates of CH<sub>4</sub>, while optimum loading rates promote high bioconversion efficiencies of complex 20 organic matter into CH<sub>4</sub> (Hill and Bolte, 1989). Data in Figure 2 and Table 1 provide evidence 21 that the biological processing events occurring in high-load systems (Type 1 and Type 2) may 22 result from the functional decoupling of the anaerobic food chain. This observation is further 23 supported by the fact that these same systems show an accumulation of organic (particulate C, H, 24 N) material (Table 1) and high air concentrations of VOCs (Fig. 3; Table 3). In contrast, the high 25 emission rates for CH<sub>4</sub> and low liquid-phase organic content (particulate C, H, N) associated with 26 Type 3 and 4 systems (Table 1) indicates that the environment in these systems provides for 27 relatively more efficient bioconversion of complex organic substrates into CH<sub>4</sub>. 28 Data reported in this study were collected in three geographical regions of the U.S. over a 29 six week period in the late summer (August and September, 1997). This experimental design 30 was chosen based on previous reports that bacterial photosynthetic and SO<sub>4</sub> reduction activities

in swine manure management systems located in northern U.S. climates reach a near steady-state condition during this time period (Do et al., 1998; Do et al., 1999; Jacobson et al., 1997b). A second objective of this design was to identify differences in emission or effluent properties of swine manure management systems that could be attributed to geographical location (Iowa, n = 24; Oklahoma, n = 2; and North Carolina, n = 3). However, no significant differences were observed in any of the parameters measured in this study when "like" manure management systems (i.e., photosynthetic lagoons *versus* photosynthetic lagoons) were compared on the basis of geographic location. While the certainty of this finding is diminished by the low sample number, it was not entirely surprising that effects of geographic location played a minor role in the emission characteristics of (steady-state) manure management systems when contrasted to the effects due to loading rate. While the effects of geographic location on emission parameters appeared to be insignificant, geographic location was found to influence the rate at which the apparent steady-state condition was achieved. Based on the population dynamics of purple nonsulfur photosynthetic populations, it has been noted that that the apparent steady-state condition occurs at least 1.2 months earlier in North Carolina lagoons when compared to comparable lagoon systems in Iowa (Do et al., 1998). The seasonal transition in P and S concentrations for photosynthetic lagoons in Iowa progress from "basin-like" characteristics in early spring (100 mg L<sup>-1</sup> P and 30 mg L<sup>-1</sup> S) to intermediate concentrations ("lagoon-like") in late spring (60 mg L<sup>-1</sup> P and 16 mg L<sup>-1</sup> S), to low concentrations during the photosynthetic bloom event in late June (1 mg L<sup>-1</sup> P and 8 mg L<sup>-1</sup> S) (Do et al., 1998). Maximum methane flux during these transition events was observed to occur shortly (2-9 days) before the photosynthetic bloom (Do et al., 1998). These studies indicate that the P and S concentrations may provide valuable swine manure management classification information regardless of the season or geographical location.

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#### Relationships between Emission Chemistry and Odor Intensity

Several recent investigations have attempted to define relationships between chemical concentration of specific gases and odor concentration or intensity (Hobbs et al. 1995, Jacobson et al. 1997a, Jacobson et al. 1997b, Obrock-Hegel 1997, Pain et al. 1990). Obrock-Hegel (1997), found that nutritional manipulation of amino acid intake reduced NH<sub>3</sub>, cresols, and indoles measured in air samples from production environments. However, no reduction in odor

1 concentration was observed between control and treatment samples. Schulte et al. (1985) and 2 Hobbs et al. (1995), linked high levels of ammonia (NH<sub>3</sub>) to odor. Unfortunately, the latter 3 authors noted that the relationship between NH<sub>3</sub> and odor could not be universally applied to all 4 farms, especially when they differed in the type of manure management system utilized. The use 5 of hydrogen sulfide (H<sub>2</sub>S) as a surrogate of livestock manure odor has also proven to be a 6 formidable challenge. Jacobson et al., (1997a) evaluated odor and H<sub>2</sub>S concentration in air from 7 approximately 60 different pig, dairy, beef, and poultry manure storage units on farms in 8 Minnesota. Low correlation was observed between H<sub>2</sub>S and odor concentration for manure 9 storages based on a species comparison and for production systems grouped according to manure 10 management system type (pit, basin, and lagoon). The study further suggested the possibility that 11 chemical odorants other than H<sub>2</sub>S (i.e., VOCs) were responsible for swine odor. In support of 12 this conclusion, Powers et al., (1999) recently demonstrated that effluent concentrations of 13 several VOCs, present in anaerobic digester effluent were correlated with odor intensity. 14 However, effluent concentration of VOC did not predict odor intensities well enough to suggest 15 that human panels should be eliminated. Data quality in this study was likely influenced by the 16 fact that correlations were not performed in a similar matrix (air versus liquid). Previous studies 17 have established that effluent measurements often provide an inaccurate representation of 18 malodor potential and therefore, represent an inappropriate comparison (Zahn et al., 1997). 19 To test the potential relationship between airborne VOCs and swine manure odor, odor 20 intensity and air concentration of VOCs were determined simultaneously at the 29 swine 21 production facilities. Figure 2C shows the relationship between average odor intensity, assessed 22 through direct scaling techniques using a defined odor intensity, and the total air concentration of 23 VOCs present at the receptor. The relationship between mean odor intensity and air 24 concentration of VOCs was observed to obey Michaelis-Menten kinetics, with a Michaelis constant (K<sub>m</sub>) of 95 µg m<sup>-3</sup> VOCs and receptor saturation occurring at 1250 µg m<sup>-3</sup> VOCs (Fig. 25 26 2). The saturation properties showed typical second-order kinetics and were characteristic of a 27 receptor-ligand process. A qualitative analysis of VOCs present at sites representing the four 28 classes of swine manure management systems is shown Figure 3 and Table 3. The gas 29 chromatography-flame ionization detector (GC-FID) chromatograms from deep pits or basins 30 produced high odor intensities and relatively intense signal responses but were chemically-

1 simplistic in nature when compared to chromatograms from lagoons (Fig. 3; Table 3). These 2 results provide evidence that chemical concentration rather than diversity is the largest 3 determinate in the odor intensity response associated with swine manure odor. This observation 4 is important since the ability to define odorant synergisms and antagonisms has been suggested to 5 be the most significant obstacle in applying chemical methods in odor measurement (Mackie et 6 al., 1998). Data presented in Figure 2C provide evidence that the total air concentration of non-7 methane VOCs predict the odor intensity associated with swine manure management systems 8 evaluated in this study. 9 Odor measurement methods using human olfactory senses are based on the use of 10 psychophysical theory. As the name implies, psychophysical methods are based on relationships 11 between psychological and physical attributes of sensory stimuli. The intensity of olfactory 12 stimuli reported by an individual is related to stimulus magnitude. For many odorants used in the 13 food and fragrance industry, there is a linear relationship between log olfactory intensity reported 14 by the individual and the air concentration of the odorant(s) present in air (Turk and Hyman, 15 1991). This relationship between perceived olfactory stimuli and intensity of sensation is 16 referred to as the fundamental psychophysical law (Stevens, 1957; Stevens, 1962). Data reported 17 in Figure 2C show that the total air concentration of VOCs correlate well to the log stimulus 18 intensity ( $r^2 = 0.88$ ) and therefore, conforms to the fundamental psychophysical law. In 19 agreement with other olfactory studies, we observed that data quality was influenced by variables 20 associated with the subjective nature of intensity scales, fatigue, sex, age, race, and visual cues 21 (Cain et al., 1998; Degel and Koster, 1998; Liden et al., 1998; Livermore and Laing, 1998; Turk 22 and Hyman, 1991). However, the use of the defined odor standard Z-2 was found to reduce most 23 of the sampling variability associated with intensity scales. Evidence for this conclusion was 24 provided by the low average standard deviation reported by panelists for site evaluations (avg. 25 std. dev. = 0.14 odor units). A prerequisite that underlies all olfactory methods is the ability to 26 define and properly sample chemical odorants that constitute a particular odor. Accomplishing 27 this objective has been a formidable challenge, since it requires the use of both olfactory and 28 analytical methods to validate sampling methods. The results of this study demonstrate utility of 29 chemical methods in odor analysis and in the validation of air sample collection methods (i.e., 30 Teflon bag sampling).

#### **Emission Rate of Air Pollutants from Swine Manure Management Systems**

Analysis of CH<sub>4</sub> emission rates, airborne VOC concentration, and odor intensity for the 29 swine manure management systems evaluated in this study indicated that manure management systems could be functionally classified according to the ratio of P to S concentrations. This analysis, however, provided limited insight into the differences in microbial processes associated with these systems. In an effort to further elucidate chemical and biological differences associated with the system classification, a detailed evaluation of effluent characteristics and gases (H<sub>2</sub>S, NH<sub>3</sub>, CH<sub>4</sub>, VOCs) was conducted at four swine manure management systems that represented each of the four system classes. Each of the swine manure management systems chosen for the follow-up study were shown to exhibit the same relative profiles in CH<sub>4</sub> emission rate regardless of the flux measurement strategy employed (Table 1, chamber-based vs. Table 4, micrometeorological-based). These results provided additional support that manure management loading parameters influenced bioconversion efficiency. In systems with relatively low emission rates of CH<sub>4</sub> and NH<sub>3</sub> (Type 1 and 2 systems, Tables 3, 4, and 5), high concentrations of organic carbon and nitrogen (particulate C, H, N) were found to accumulate in the solution phase (Table 1). These systems were also observed to have the highest odor intensities. In contrast, Type 3 and 4 systems showed a lower tendency to accumulate effluent organic material (Fig. 3; Tables 1 and 3) and a much higher emission rate of CH<sub>4</sub> (Fig. 2; Table 1).

The emission rate of hydrogen sulfide (H<sub>2</sub>S) appeared to be independent of the manure management system classification developed in this study (Table 5). This observation was unexpected since the emission rate of CH<sub>4</sub>, NH<sub>3</sub>, and VOCs were dependent upon the type of manure management system utilized (Fig. 2.; Table 5). Common precursors of H<sub>2</sub>S in anaerobic swine manure management systems may include cysteine (cystine), methionine (indirect biotransformation), sulfur (S<sub>0</sub>), thiosulfate (S<sub>2</sub>O<sub>3</sub>), or sulfate (SO<sub>4</sub>). Sulfate (SO<sub>4</sub>) is known to play a major role in mammalian physiology and is released (~0.1 to 1.4 g of SO<sub>4</sub> day<sup>-1</sup>) in urinary excretions as SO<sub>4</sub>-organic conjugates or SO<sub>4</sub> salts (Lehninger, 1988). Based on a daily SO<sub>4</sub> excretion rate of 0.13 g SO<sub>4</sub> pig<sup>-1</sup> day<sup>-1</sup> and a complete conversion to H<sub>2</sub>S, a production facility with 4560 pigs (Site A, Table 5) is estimated to emit 593 g H<sub>2</sub>S day<sup>-1</sup> through dissimilatory SO<sub>4</sub> reduction processes (Gottschalk, 1988; Postgate, 1984). This calculated value is similar to the

1 observed daily H<sub>2</sub>S emission rate (Table 5, Site A), indicating that urinary SO<sub>4</sub> excretions may 2 account for a significant proportion of the sulfur (S)-precursors contributing to the H<sub>2</sub>S emissions 3 measured in this study. The finding that SO<sub>4</sub> may be the major S-precursor contributing to H<sub>2</sub>S 4 emissions provides microbiological insight into why H<sub>2</sub>S emissions might be independent of 5 manure management system classification. All complex, organic forms of sulfur (S) require 6 energy expenditure to produce volatile, S-containing gases (i.e., formation of methyl mercaptan 7 or dimethyl disulfide from methionine; Gottschalk, 1988). Energy consuming microbial 8 processes have been shown to be highly sensitive to effluent environmental parameters such as 9 metal ion concentration and concentration of organic matter (Oleszkiewicz and Sharma, 1990; 10 Hill and Bolte, 1989). In contrast, SO<sub>4</sub> reduction is an energy-yielding process that is highly 11 favorable under environmental conditions observed in all manure management systems evaluated 12 in this study. These results suggest that future mass-balance studies should aim at 13 characterization of the S-cycle in swine manure management systems with special emphasis on 14 sources and fate of SO<sub>4</sub>. 15 Manure management systems with high loading rates and relatively low CH<sub>4</sub>-producing 16 activity showed up to a 26-fold increase in total VOC emissions when compared to low-odor 17 photosynthetic lagoons (Fig. 2 and Table 5). The data indicate that VOCs are more likely to be 18 of concern with systems employing high loading rates, while CH<sub>4</sub> and NH<sub>3</sub> are likely to be more 19 problematic with systems employing lower loading rates. In addition to nuisance concerns, 20 elevated VOC concentrations may present a concern to human health. A small number of 21 regulated industrial pollutants are present in airborne emission streams from swine production 22 facilities (Table 3). The air concentrations for these compounds at the source were found to be at 23 least one order of magnitude below exposure levels established for safe work environments by 24 occupational health organizations (Plog, 1988). However, the use of industrial exposure indices 25 may not be appropriate for assessing exposure to animal waste emissions due to the following 26 considerations: First, several agricultural waste pollutants are unique to agricultural systems and 27 thus, have no established exposure indices. Second, bioactive airborne pollutants, such as 28 microbial secondary metabolites, remain largely uncharacterized from swine production systems. 29 Bioactive compounds have been identified as a serious human health risk, since many of these 30 compounds have been shown to influence mammalian cell physiology in the part per billion and

1 sub-part per billion range (Andersson et al., 1998). Third, there is a lack of information on the 2 acute and chronic toxicological impacts of VOCs derived from swine manure on children and 3 individuals of compromised health. This concern is reflected in recent epidemiological studies 4 that have shown a higher incidence of psychological dysfunction and health-related problems in 5 individuals residing near large-scale swine production facilities (Thu et al., 1997; Schiffman et 6 al., 1995). 7 Ammonia emissions from the four intensively studied swine production systems were found 8 to violate release reporting requirements for NH<sub>3</sub> under the U.S. EPA Comprehensive 9 Environmental Response, Compensation, and Liability Act (CERCLA, U.S. EPA, 2000). 10 Emergency Planning and Community Right-to-Know Act (EPCRA) section 329(4), defines 11 facility to include stationary structures on a single site, or on contiguous or adjacent sites owned 12 or operated by the same person. Under this definition, the aggregated emission rate of registered 13 hazardous substances (i.e., NH<sub>3</sub>, H<sub>2</sub>S, VOC, particulate matter) from all swine production facility 14 point sources is subject to release reporting requirements. The current reporting requirements for NH<sub>3</sub> (and H<sub>2</sub>S) are set at 100 pounds of NH<sub>3</sub> day<sup>-1</sup> (CERCLA, U.S. EPA, 2000). The range for 15 NH<sub>3</sub> emissions from the four intensively studied swine production sites ranged from 101.7 kg 16 NH<sub>3</sub> day<sup>-1</sup> (224 lbs. NH<sub>3</sub> day<sup>-1</sup>) to 369.2 kg NH<sub>3</sub> day<sup>-1</sup> (813.9 lbs. NH<sub>3</sub> day<sup>-1</sup>). The observed 17 18 aggregate emission rates for swine production facilities evaluated in this study exceed the 19 CERCLA reporting requirements for NH<sub>3</sub> by 55% to 88%. Under section 304 of EPCRA, the 20 "owner or operator" of a facility is required to report immediately to the appropriate State 21 emergency response commissions and local emergency planning committees when there is a 22 release of a CERCLA hazardous substance. The results indicate that NH<sub>3</sub> emissions from swine 23 production facilities have the potential to exceed release reporting requirements enforced by the 24 U.S. EPA. 25 Lagoons with established anoxic bacterial photosynthetic populations (Type 4 systems; 26 bacteriochlorophyll a concentrations above 40 nmol/ml) showed lower odor intensities, lower air 27 concentrations of VOCs, and lower emission rates of VOCs when compared to other swine 28 manure management systems (Fig. 3; Tables 3 and 4). Photosynthetic bacteria carry out the 29 process of photosynthesis under anaerobic conditions. These requirements for photosynthesis 30 differ greatly from algae and plant species that use water as an electron source for photosystem II

and evolve oxygen in this reaction (Kobayashi and Kobayashi, 1995). Instead of utilizing water as a reductant, anoxic photosynthesis is dependent on substrates such as H<sub>2</sub>S, hydrogen (H<sub>2</sub>), and VOCs to provide reducing equivalents, while light from the sun provides the energy source (Gottschalk, 1988). Bacteria capable of anoxic photosynthesis are categorized into one of four major groups (purple sulfur, purple non-sulfur, green sulfur, and green-gliding) based on the presence of specific types of bacteriochlorophyl, phylogenetic characteristics, and metabolic capabilities (Gottschalk, 1988; Siefert et al., 1978). All four groups of photosynthetic bacteria are able to utilize organic substrates (VOCs) as a source of carbon and the green-gliding and purple bacteria are able to use organic substrates as proton donors. However, a novel species of *Rhodobacter (Rhodobacter* sp. PS9) dominates the photosynthetic population (~20% of the total microbial community structure) in all photosynthetic swine lagoon systems examined in this study (Do et al., 1998; Do et al., 1999). The physiological characteristics of this purple non-sulfur photosynthetic bacterium provides evidence for the observed degradation of VOCs and decreased odor emissions from photosynthetic swine waste lagoons.

#### 16 CONCLUSIONS

The lack of knowledge concerning functional aspects of swine manure management systems has complicated present efforts to develop or improve emission abatement technologies to meet the stringent air quality and nuisance regulations presently imposed by several States. The task of elucidating functional aspects of swine manure management systems is often convoluted by the plethora of manure management strategies presently used in the swine industry. In the absence of appropriate compatibility or functional measures, emission abatement strategies are often paired with swine manure management systems based only on empirical observations. As a result, the performance of these abatement methods is difficult to predict and often impossible to improve through scientific methods. Effluent concentrations of P and S are shown in this study to be a useful tool in the classification of swine manure management systems according to functional aspects of the system. The methods described in this study provide an inexpensive means to rapidly assess best management practices for swine manure management systems and also provide a means by which to identify swine production systems that represent a potential air quality or nuisance concern.

1	
2	ACKNOWLEDGEMENTS
3	Research support from the National Pork Producers Council (to J.A. Zahn, A.A.
4	DiSpirito, and J.L. Hatfield), Iowa State University (I.S.U.) Office of Biotechnology (to A.A.
5	DiSpirito), and from the USDA-ARS ( to J.A. Zahn, and J.L. Hatfield) is gratefully
6	acknowledged. The assistance of Mr. Tim Hart (National Soil Tilth Laboratory (NSTL)) with
7	data collection, Ms. Lori Merrill (I.S.U.) with manure dry weight analysis, and Mr. Pierce
8	Fleming (NSTL) and Dr. David Laird (NSTL) with ICP-AES analysis is kindly acknowledged.
9	The informative discussions with Dr. John Prueger (NSTL) concerning application of
10	meteorological approaches in measurement of evaporative fluxes is also acknowledged. A
11	special thanks to the following site collaborators for their valuable contributions in site selection
12	and for allowing access to sites: Dr. Thomas Sauer (USDA Biomass Res. Ctr., Univ. AR), Mr.
13	James Seward and Joe Locke (Tyson Foods, Inc., Springdale, AR), Mr. Brent Rastetter
14	(Heartland Pork, Alden, IA), Mr. Alan Witt (Murphy Family Farms, Algona, IA), Dr. Matias
15	Vanotti and Dr. Patrick Hunt (USDA Costal Plains Soil, Water, and Plant Res. Ctr., Florence,
16	SC), Mr. Brent Roberts (Monsanto, EnviroChem Systems Division, St. Louis, MO), and Mr.
17	Alan Junck, (Junck Farms, Ogden, IA).
18	
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**TABLE 1.** Physical properties, elemental composition, and  $CH_4$  emission rates determined by flux chamber methods for 29 swine manure management systems located in Iowa, Oklahoma, and North Carolina during the months of August and September, 1997. Values represent the mean  $\pm$  the standard error of the mean.

Parameter †	Site Classification ‡			
_	Type 1 ‡	Type 2	Type 3	Type 4
Site number (n)	n = 6	n = 7	n = 6	n = 10
Manure management	DP; $n = 5$	EB; $n = 3$	L; $n = 6$	PL; n = 10
system description	PP; $n = 1$	CLB; $n = 3$		
		ST; n = 1		
Methane emission rate	$636 \pm 47$	$1830 \pm 148$	$13900 \pm 760$	$11990 \pm 540$
(g CH <sub>4</sub> system <sup>-1</sup> hr <sup>-1</sup> )				
Volatile solids loading rate	$79 \pm 3.0$	$35 \pm 2.6$	$0.3 \pm 0.05$	$0.07 \pm 0.02$
(kg VS day <sup>-1</sup> m <sup>-3</sup> )				
pН	$7.1 \pm 0.04$	$7.3 \pm 0.06$	$7.3 \pm 0.06$	$7.1 \pm 0.03$
Solid content (mg/ml)	$21.9 \pm 0.9$	$13.4 \pm 0.6$	$3.8 \pm 0.4$	$2.8 \pm 0.1$
% Carbon (% dry mass)	$37.2 \pm 0.6$	$33.7 \pm 0.5$	$16.6 \pm 0.3$	$14.1 \pm 0.3$
% Hydrogen (% dry mass)	$5.2 \pm 0.3$	$4.9 \pm 0.2$	$2.3 \pm 0.2$	$1.8 \pm 0.1$
% Nitrogen (% dry mass)	$3.0 \pm 0.2$	$2.9 \pm 0.2$	$1.8 \pm 0.2$	$1.5 \pm 0.1$
Ca (mg/L)	$301 \pm 38$	$173 \pm 24$	$58 \pm 5$	$119 \pm 10$
Cu (mg/L)	$18 \pm 6.2$	$1.7 \pm 0.2$	$0.2 \pm 0.1$	$0.2 \pm 0$
Fe (mg/L)	$47 \pm 21.5$	$8.9 \pm 1.2$	$1.2 \pm 0.2$	$0.7 \pm 0.4$
K (mg/L)	$1380 \pm 400$	$1040 \pm 134$	$624 \pm 18$	$0.9 \pm 0.3$
Mg (mg/L)	$128 \pm 19$	$62 \pm 10$	$20 \pm 1.6$	$39 \pm 4.0$
Mn (mg/L)	$3.2 \pm 0.8$	$0.9 \pm 0.2$	$0.1 \pm 0$	$0.3 \pm 0.2$
Na (mg/L)	$241 \pm 86$	$225 \pm 20.8$	$165 \pm 4.6$	$18 \pm 6.3$
P (mg/L)	$504 \pm 26$	$153 \pm 12.1$	$65 \pm 4.5$	$0.2 \pm 0$
S (mg/L)	$108 \pm 8$	$39 \pm 5.3$	$15 \pm 0.4$	$8 \pm 1.8$
Zn (mg/L)	$18.7 \pm 8$	$2.7 \pm 0.4$	$0.4 \pm 0.1$	$0.1 \pm 0$

 $<sup>\</sup>dagger$  Values represent the mean for samples listed in Fig. 1. Instrumental error was < 1% for CHN analysis and < 0.1% for ICP-AES analysis.

<sup>‡</sup> Manure storage system designation as defined in Fig. 1. Type 1 systems represent confinement buildings with short and long term, under-slat storage (pull-plug and deep-pit systems). Type 2 systems represent earthen, concrete, or steel-lined manure storage basins. Type 3 and Type 4 systems represent lagoon systems without and with anoxic photosynthetic blooms, respectively. Subclassification designations: PL = phototrophic lagoon; L = lagoon; EB = earthen basin; CLB = concrete-lined basin (outdoor); ST = steel tank (outdoor); DP = deep pit; PP = pull-plug.

**TABLE 2.** Results for the pairwise comparison of element concentrations present in effluent samples from 29 swine manure management systems. Values indicate the nonparametric measure of association between pairwise comparisons of elements.

	Nonparametric measure of association (Spearmans rho value)									
Element	Ca	Cu	Fe	K	Mg	Mn	Na	P	S	Zn
Ca	1	0.65	0.80	0.92	0.82	0.37	0.66	0.66	0.66	0.66
Cu		1	0.92	0.70	0.66	0.83	0.65	0.96	0.92	0.96
Fe			1	0.64	0.78	0.95	0.59	0.93	0.89	0.91
K				1	0.36	0.56	0.96	0.74	0.74	0.67
Mg					1	0.83	0.30	0.69	0.68	0.66
Mn						1	0.50	0.87	0.85	0.84
Na							1	0.67	0.66	0.62
P								1	0.94	0.96
S								•	1	0.93
Zn										1

**TABLE 3.** The mean concentration of trace gases and VOCs present in air samples from four swine manure management systems representing each system class. Values reported represent the sample mean for a 24 hr sampling period and the standard error was less than 3% of the mean. Sample number site<sup>-1</sup>: VOC, n = 12;  $NH_3$ ,  $CH_4$ , and  $H_2S$ , n = 24.

Peak # and Compound <u>Swine manure management system type</u>							
Retention	•		nalyte air concentration (µg m <sup>-3</sup> ) and Percent				
time		total peak area (%))					
(min.)		Type 1 †	Type 2	Type 3	Type 4		
Spec	Ammonia	9623	7923	9362	10843		
GC-FID	Methane	5002	8406	18703	24406		
GC-FPD	Hydrogen Sulfide ‡	54	48	27	29		
1 (4.8)	Dimethyl disulfide	12 (1.3)	nd	17 (6.8)	nd		
2 (6.1)	2-Butanol	8 (0.8)	nd	19 (7.5)	nd		
3 (7.5)	Dimethyl trisulfide	nd	nd	13 (5.2)	nd		
4 (8.6)	Unknown	nd	nd	- (2.9)	nd		
5 (10.6)	Acetic acid	281 (15.2)	262 (7.6)	11 (2.7)	2 (2.3)		
6 (11.9)	Propionic acid	126 (11.1)	50 (2.3)	5 (1.9)	4 (8.2)		
7 (12.4)	Isobutyric acid	23 (2.5)	107 (11.4)	6 (2.2)	nd		
8 (13.3)	Butyric acid	142 (15)	586 (32)	13 (5.1)	5 (12.9)		
9 (14.0)	Isovaleric acid	73 (8.3)	98 (6)	3 (1.2)	nd		
10 (15.0)	n-Valeric acid	43 (4.9)	360 (27)	5 (2.0)	1 (0.7)		
11 (15.7)	Isocaproic acid	nd	10 (0.5)	nd	nd		
12 (16.0)	n-Caproic acid	nd	105 (7.4)	nd	nd		
13 (16.1)	Unknown	- (1.2)	nd	nd	nd		
14 (16.6)	Heptanoic acid	nd	8 (0.3)	nd	nd		
15 (17.2)	Benzyl alcohol	nd	nd	2 (1.2)	nd		
16 (18.8)	Phenol	9 (1.5)	24 (1.6)	8 (6.9)	3 (9.9)		
17 (19.7)	4-Methyl phenol ‡	85 (19.6)	32 (2.7)	9 (7.5)	3 (17.8)		
18 (20.9)	4-Ethyl phenol ‡	3 (0.7)	2 (0.2)	4 (3.3)	1 (6)		
19 (21.9)	2-Amino	nd	nd	nd	0.2 (0.4)		
	acetophenone ‡						
20 (23.4)	Indole	nd	1.1 (0.2)	0.8(0.6)	0.1 (0.5)		
21 (23.7)	Hexadecanoic acid	nd	nd	9 (7.8)	5 (33)		
22 (24.2)	3-Methyl indole	0.5 (0.2)	1.4 (0.3)	1.3 (1.1)	0.2 (0.7)		
Total co	ncentration of non-	806	1647	126	25		
methane V	OCs identified in air						
$(\mu g m^{-3})$							
Percent of	f total peak area (%)	82.3	99.5	65.9	92.4		

<sup>†</sup> Micrometeorologically-defined sampling parameters: Type 1 = mechanically-ventilated exhaust flow rate = 110,000 m<sup>3</sup> hr<sup>-1</sup>, sampling position at the fan orifice; Type 2 = 39 m diameter basin, z=ZINST(0.1 cm roughness length, 1950 cm radius = 70 cm sampling height); Type 3 = 92 m diameter lagoon, z=ZINST(0.1 cm roughness length, 4600 cm radius = 189 cm

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sampling height); Type 4 = a primary 100 m diameter photosynthetic lagoon, z=ZINST(0.1 cm roughness length, 5000 cm radius = 191 cm sampling height).

U.S. EPA priority pollutants identified in air samples: cresols (isomers and mixtures), H<sub>2</sub>S, phenol, and acetophenones. nd = analyte not detected.
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**TABLE 4.** Mean horizontal flux rate and mean emission rate for analytes collected over a 24 hr sampling period from swine manure management systems sampled in Table 3 and described in Table 5. Gas flux rates were measured using the Theoretical Profile Shape micrometeorological method.

Parameter	Swine manure management system type					
	Type 1	Type 1 Type 2 Type 3 Typ				
Mean wind velocity <sup>a</sup> (cm s <sup>-1</sup> ) or ventilation rate <sup>b</sup> (m <sup>3</sup> hr <sup>-1</sup> ) during sampling period	110,000 <sup>b</sup>	190ª	128ª	90ª		
Ammonia flux rate (ng NH <sub>3</sub> cm <sup>-2</sup> s <sup>-1</sup> )	66†	167	109	89		
Ammonia emission rate (g NH <sub>3</sub> system <sup>-1</sup> hr <sup>-1</sup> )	1060	1900	7700	6270		
Methane flux rate (ng CH <sub>4</sub> cm <sup>-2</sup> s <sup>-1</sup> )	34 †	178	218	200		
Methane emission rate (g CH <sub>4</sub> system <sup>-1</sup> hr <sup>-1</sup> )	550	2010	15410	14120		
Hydrogen sulfide flux rate (ng H <sub>2</sub> S cm <sup>-2</sup> s <sup>-1</sup> )	0.37 †	1.10	0.32	0.24		
Hydrogen sulfide emission rate (g H <sub>2</sub> S system <sup>-1</sup> hr <sup>-1</sup> )	5.9	12.5	22.7	16.9		
Priority pollutant (PP) flux rate (ng PP cm <sup>-2</sup> s <sup>-1</sup> ) ‡	1.04 †	2.30	0.56	0.30		
Priority pollutant emission rate (g PP system <sup>-1</sup> hr <sup>-1</sup> ) ‡	16.6	26.1	39.6	20.9		
VOC flux rate (ng VOC cm <sup>-2</sup> s <sup>-1</sup> )	5.60 †	35.0	1.60	0.21		
VOC emission rate (g VOC system <sup>-1</sup> hr <sup>-1</sup> )	89.9	394.0	113.1	14.5		
Total air pollutant emission rate (g TAP system <sup>-1</sup> hr <sup>-1</sup> )	1720	2420	15550	14150		

<sup>†</sup> System flux rate calculated using an active surface area of 4,459,000 cm<sup>2</sup> and assumes a homogenous emitting source for active surfaces.

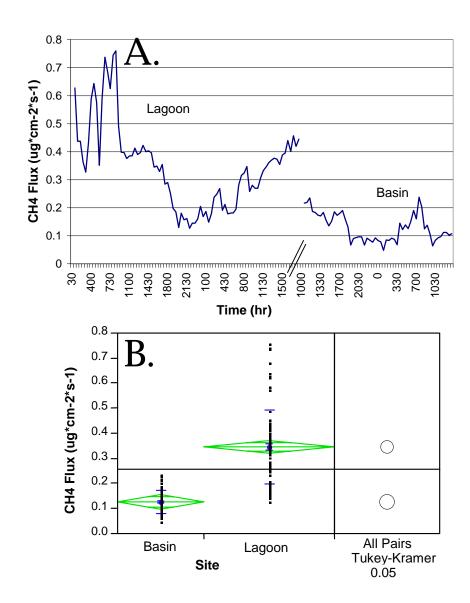
<sup>‡</sup> U.S.-EPA priority pollutants identified in air samples: cresols (isomers and mixtures), hydrogen sulfide, phenol, and acetophenones.

**TABLE 5.** Emission rates for air pollutants collected over a 24 hr sampling period from swine manure management systems sampled in Tables 3 and 4. Values were calculated from flux measurements that employed the Theoretical Profile Shape micrometeorological method.

Parameter	Swine manure management system type				
	Type 1‡	Type 2	Type 3	Type 4	
Annual production number and	Feeder to	Farrow to	Feeder to	Farrow to	
type	finish	finish	finish	feeder	
	13680	8200	14170	18500	
	animals	animals	animals	animals	
	yr <sup>-1</sup>	yr <sup>-1</sup>	yr <sup>-1</sup>	yr <sup>-1</sup>	
Site manure management system	4 - DP	2 - CLB	1 - L	2 - PL	
description †		2 - PP	PFS ‡	PFS ‡	
Ammonia emission	101.7	141.7	232.8	369.2	
(kg NH <sub>3</sub> site <sup>-1</sup> day <sup>-1</sup> )					
Methane emission	52.8	122.7	466.1	831.0	
(kg CH <sub>4</sub> site <sup>-1</sup> day <sup>-1</sup> )					
Hydrogen sulfide emission	0.6	0.9	0.7	1.0	
$(kg H_2 S site^{-1} day^{-1})$					
Priority pollutant (PP) emission	1.6	2.1	1.2	1.2	
(kg PP site <sup>-1</sup> day <sup>-1</sup> )					
VOC emission	8.6	23.2	3.4	0.9	
(kg VOC site <sup>-1</sup> day <sup>-1</sup> )					
Combined air pollutant emission	165	291	704	1203	
(kg TAP site <sup>-1</sup> day <sup>-1</sup> )					

<sup>†</sup> Manure storage system description: PL = phototrophic lagoon; PFS = continuous pit flush system; L = lagoon; EB = earthen basin; CLB = concrete-lined basin (outdoor); DP = deep pit; PP = pull-plug.

<sup>‡</sup> The emission rate for analytes released from the continuous pit flush systems was calculated based the flux rates (Table 4) and the total pit surface area actively flushed with recycled lagoon liquid. The active surface area for PFS systems at site 3 and 4 was 510 m² and 890 m², respectively.



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**FIGURE 3.** Gas chromatograms of volatile organic emissions from swine manure storage systems representing the four system subtypes. From top to bottom: Air sample collected in a deep pit (Type 1), air sample collected over a concrete-lined basin (Type 2), air sample collected over a lagoon (Type 3), and air sample collected over a photosynthetic lagoon (Type 4). Peak reference numbers correspond to organic compounds listed in Table 3.

